Isolation of DNA from bacterial, plant and animal cells

Aim:

To isolate DNA from bacterial, plant, and animal cells for subsequent analysis.

Principle:

The principle behind DNA isolation involves breaking open the cells to release the DNA, separating it from other cellular components such as proteins, membranes, and RNA, and then precipitating the DNA out of solution.

Materials and Methodology:

Materials:

- Bacterial cells (e.g., Escherichia coli)
- Plant tissue (e.g., spinach leaves)
- Animal tissue (e.g., chicken liver)
- Buffer solutions (e.g., Tris-EDTA buffer, lysis buffer)
- Enzymes (e.g., lysozyme for bacterial cells)
- Proteinase K (for digesting proteins)
- Phenol:chloroform

alcohol (for extraction)

- Ethanol (for precipitation)
- RNase (to degrade RNA)

Methodology:

- 1. Cell Lysis:
 - **Bacterial Cells**: Treat with lysozyme to weaken cell walls, followed by detergent treatment to disrupt membranes.
 - **Plant Cells**: Grind plant tissue with buffer and enzymes to break cell walls and release DNA.
 - **Animal Cells**: Homogenize tissue with buffer and enzymes to disrupt cell membranes and release DNA.
- 2. Protein Removal:
 - Add proteinase K to digest proteins that could interfere with DNA extraction.
- 3. **DNA Extraction**:
 - Extract DNA using phenol:chloroform

alcohol mixture to separate DNA from proteins and other cellular debris.

4. Precipitation:

• Precipitate DNA by adding cold ethanol, which causes DNA to come out of solution.

5 Purification:

 \circ $\,$ Wash DNA pellet with ethanol to remove residual salts and contaminants.

6. Analysis:

• Assess the quantity and quality of isolated DNA using spectrophotometry or gel electrophoresis.

Procedure

- 1. **Preparation**:
 - Collect cells or tissue samples.
 - Prepare appropriate buffers and solutions.
- 2. Cell Lysis:
 - Treat cells/tissue with lysis buffer and appropriate enzymes.

3. Protein Digestion:

 \circ $\;$ Add proteinase K to digest proteins.

4. **DNA Extraction**:

• Extract DNA using phenol:chloroform

alcohol.

5. Precipitation:

- Precipitate DNA with cold ethanol.
- 6. Washing and Purification:
 - Wash DNA pellet with ethanol to remove contaminants.

7. Quantification:

• Measure DNA concentration and purity.

Expected Results:

- You should obtain purified DNA from bacterial, plant, and animal cells.
- The concentration of DNA can vary depending on the starting material and the efficiency of extraction.
- The purity of DNA can be assessed by measuring absorbance ratios (A260/A280) or by running an agarose gel electrophoresis.

This protocol allows for the isolation of DNA from different types of cells, which can then be used for various downstream applications such as PCR, sequencing, or genetic analysis.

Isolation of RNA from yeast cells.

Aim:

The aim of this experiment is to isolate RNA from yeast cells. RNA isolation is crucial for studying gene expression, RNA processing, and other molecular biology applications.

Principle:

RNA isolation involves breaking open yeast cells to release RNA while preserving its integrity and purity. This process typically involves disrupting the cells, separating RNA from DNA and proteins, and purifying it for downstream applications.

Materials and Methodology:

Materials:

- Yeast cells (Saccharomyces cerevisiae)
- Trizol or similar RNA extraction reagent
- Chloroform
- Isopropanol
- Ethanol
- RNase-free water
- RNase inhibitors
- Centrifuge tubes
- Microcentrifuge
- Spectrophotometer or fluorometer for RNA quantification

Methodology:

1. Cell Lysis:

- Harvest yeast cells by centrifugation.
- Resuspend the pellet in Trizol or a similar RNA extraction reagent.
- Disrupt cells using mechanical methods (such as bead beating) or enzymatic methods (lysozyme treatment for cell wall digestion in yeast).

2. RNA Extraction:

- Add chloroform to the lysate to separate RNA from DNA and proteins.
- \circ $\,$ Centrifuge to separate the phases (aqueous, interphase, organic).
- \circ Transfer the aqueous phase containing RNA to a new tube.

3. RNA Precipitation:

- Precipitate RNA by adding isopropanol.
- \circ Incubate at room temperature or -20°C to facilitate precipitation.
- Centrifuge to pellet RNA.

4. RNA Wash:

- Wash RNA pellet with ethanol to remove residual salts and contaminants.
- Air-dry or centrifuge briefly to remove ethanol.

5.RNA Resuspension:

- Resuspend RNA pellet in RNase-free water or buffer.
- Optionally, treat with RNase inhibitors to prevent RNA degradation.

6.RNA Quantification and Quality Check:

- Measure RNA concentration using a spectrophotometer (A260 measurement) or fluorometer.
- Assess RNA purity (A260/A280 ratio) and integrity (by gel electrophoresis or bioanalyzer).

Procedure:

1. Yeast Cell Harvest:

- Grow yeast cells to mid-log phase.
- Pellet cells by centrifugation.

2. Cell Lysis and RNA Extraction:

- Resuspend cell pellet in Trizol.
- Disrupt cells using bead beating or enzymatic digestion.
- Add chloroform, mix, and centrifuge to separate phases.

3. RNA Precipitation and Wash:

- Transfer aqueous phase to a new tube.
- Precipitate RNA with isopropanol, centrifuge, and wash with ethanol.

4. RNA Resuspension:

• Resuspend RNA pellet in RNase-free water or buffer.

5. RNA Quantification:

• Measure RNA concentration and purity.

Expected Result:

The expected result of this experiment is a yield of RNA with high purity and integrity suitable for downstream applications such as RT-PCR, RNA sequencing, or Northern blotting. The RNA should have a high A260/A280 ratio (indicating minimal protein contamination) and show intact ribosomal RNA bands upon gel electrophoresis.

This protocol provides a basic outline for isolating RNA from yeast cells. Variations may exist based on specific RNA extraction kits or additional steps required for particular downstream applications. Adjustments in reagent volumes, incubation times, and temperatures may be necessary depending on the specific experimental conditions and the quantity of starting

Estimation of DNA and RNA by UV absorption method and determination of purity of nucleic acids.

Aim: The aim of this experiment is to estimate the concentration of DNA and RNA in a sample using UV absorption spectroscopy, and to assess the purity of these nucleic acids.

Principle: UV absorption spectroscopy is based on the principle that nucleic acids (DNA and RNA) absorb UV light at specific wavelengths, primarily 260 nm. The amount of UV light absorbed is directly proportional to the concentration of nucleic acids present in the sample. The ratio of absorbance at 260 nm to that at 280 nm can indicate the purity of the nucleic acids, with a higher ratio indicating purer nucleic acids (less contamination by proteins).

Materials and Methodology:

Materials:

- DNA or RNA sample
- UV spectrophotometer
- Quartz cuvettes (UV-transparent)
- Buffer solution (typically TE buffer for DNA, and Tris-EDTA for RNA)
- Distilled water (for blank measurement)
- Calculator

Methodology:

1. Preparation:

- Prepare the sample: Dilute the DNA or RNA sample in the appropriate buffer solution to obtain a measurable concentration (usually between 2-50 μ g/ml).
- Prepare blank: Use the buffer solution or distilled water as a blank for zeroing the spectrophotometer.

2. Measurement:

- \circ Zero the UV spectrophotometer with the blank solution.
- Measure the absorbance of the sample at 260 nm and 280 nm wavelengths.
- Record the absorbance values obtained.

3. Calculations:

• Calculate the concentration of DNA or RNA using the formula provided by your spectrophotometer's manual or the Beer-Lambert law:

 $Concentration (\mu g/ml) = Absorbance at 260 nm \times Dilution factor Absorptivity coefficient\text{Concentration (\mu g/ml)} = \frac{\text{Absorbance at 260 nm}} \\ times \text{Dilution factor}} {\text{Dilution factor}}$

coefficient}}Concentration (μ g/ml)=Absorptivity coefficientAbsorbance at 26 0 nm×Dilution factor

• Determine the purity of the nucleic acids by calculating the 260/280 absorbance ratio:

 $\label{eq:260/280} \begin{array}{l} 260/280 \mbox{ ratio} = Absorbance at 260 \mbox{ nmAbsorbance at 280 \mbox{ nm}} \\ \label{eq:260/280 ratio} = \mbox{ frac} \\ \mbox{ text} \\ Absorbance at 260 \mbox{ nm} \\ \label{eq:260/280} \\ \mbox{ ratio} = Absorbance at 280 \mbox{ nmAbsorbance at 260 \mbox{ nm}} \\ \label{eq:260/280} \end{array}$

Procedure:

- 1. Prepare the sample by diluting DNA or RNA in buffer solution.
- 2. Zero the spectrophotometer with the blank (buffer solution or distilled water).
- 3. Measure the absorbance of the sample at 260 nm and 280 nm.
- 4. Record the absorbance values and calculate the concentration and purity ratio using the formulas provided.

Given Result for Experiment:

Let's assume a hypothetical result:

- Absorbance at 260 nm = 1.0
- Absorbance at 280 nm = 0.5

Using the above values:

- Concentration calculation:
 - Assume a dilution factor of 10.
 - \circ Absorptivity coefficient = 50 for DNA (hypothetical value).
 - Concentration = $1.0 \times 1050 = 0.2$ \frac{1.0 \times 10}{50} = $0.2501.0 \times 10 = 0.2$ µg/ml
- Purity calculation:

Agarose gel for RNA, DNA, blotthegel

Aim:

To separate and analyze RNA and DNA samples using agarose gel electrophoresis and subsequently transfer them onto a membrane (blotting).

Principle:

Agarose gel electrophoresis is a widely used technique to separate nucleic acids based on their size. Smaller fragments move faster through the gel matrix, while larger fragments move slower. Once separated, the nucleic acids can be visualized using stains or transferred to a membrane (blotting) for further analysis, such as hybridization with specific probes.

Materials and Methodology:

Materials:

- Agarose
- TAE (Tris-acetate-EDTA) buffer
- DNA and RNA samples
- Gel loading dye (containing bromophenol blue or other tracking dyes)
- Ethidium bromide (for staining) or other nucleic acid stains
- UV transilluminator
- Gel electrophoresis apparatus
- Nylon or nitrocellulose membrane
- Blotting buffer (commonly SSC buffer for Northern blots)
- Hybridization probes (optional)

Methodology:

1. Preparing Agarose Gel:

- Prepare a suitable percentage agarose gel (typically 1-2% for separating DNA and RNA).
- Mix agarose powder with TAE buffer and heat to dissolve.
- Allow the agarose mixture to cool to about 50°C before pouring into the gel mold.
- Insert a gel comb to create wells for sample loading.

2. Loading and Running the Gel:

- Mix DNA or RNA samples with loading dye.
- Load the samples into the wells of the agarose gel.
- Run the gel in the electrophoresis apparatus filled with TAE buffer, applying a suitable voltage (typically 80-120V for DNA, lower for RNA to prevent degradation) until the dye front reaches the bottom of the gel.

3. Visualization of Bands:

• Stain the gel with ethidium bromide or another suitable stain.

• Visualize the separated nucleic acid bands under UV light using a transilluminator.

4. Blotting the Gel:

- \circ Cut the agarose gel according to the size of the membrane.
- Prepare the blotting setup with the membrane sandwiched between absorbent paper towels or filter papers soaked in blotting buffer (e.g., SSC buffer).

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- Apply gentle pressure to ensure good contact between the gel and the membrane.
- Allow capillary action to transfer the nucleic acids from the gel to the membrane.

5. Hybridization (Optional):

- After blotting, hybridize the membrane with specific probes if desired.
- Probe hybridization will allow for the detection of specific DNA or RNA sequences.

Expected Results:

- On the agarose gel: Visualize distinct bands representing DNA or RNA fragments of different sizes.
- On the membrane: After blotting and potentially hybridization, specific bands or signals corresponding to the probed sequences will appear.

This experiment allows for the separation, visualization, and transfer of nucleic acids onto membranes for further analysis, such as identifying specific sequences or comparing the sizes of DNA or RNA fragments.

Determination of sugar and phosphateratiosin DNA and RNA samples

Aim:

The aim of this experiment is to determine the sugar and phosphate ratios in DNA and RNA samples.

Principle:

DNA (deoxyribonucleic acid) and RNA (ribonucleic acid) are nucleic acids composed of nucleotide units. Each nucleotide consists of a sugar molecule (deoxyribose in DNA and ribose in RNA), a phosphate group, and a nitrogenous base. The backbone of both DNA and RNA is formed by sugar-phosphate bonds. The ratio of sugar to phosphate in these molecules is crucial for understanding their structure and function.

Materials and Methodology:

Materials:

- DNA sample (e.g., extracted from cells or synthesized oligonucleotide)
- RNA sample (e.g., extracted from cells or synthesized RNA)
- Reagents for nucleic acid extraction (if needed)
- Buffer solutions (for extraction and analysis)
- Spectrophotometer or other analytical equipment
- Disposable cuvettes or sample holders

Methodology:

- 1. Sample Preparation:
 - Extract DNA and RNA samples using standard protocols if necessary.
 - Ensure samples are purified to remove contaminants that could interfere with spectrophotometric analysis.

2. Spectrophotometric Analysis:

- \circ $\;$ Set up the spectrophotometer according to manufacturer instructions.
- Blank the spectrophotometer with appropriate buffers or solvents.
- Measure the absorbance of DNA and RNA samples at specific wavelengths (typically around 260 nm and 280 nm).

3. Calculation of Ratios:

- Use the absorbance values to calculate the concentration of DNA and RNA samples.
- Calculate the ratio of nucleic acid bases (e.g., A, T, G, C for DNA; A, U, G, C for RNA) to determine purity and content.
- Determine the sugar to phosphate ratios based on known molar absorptivity values and concentrations.

Procedure:

- 1. Prepare samples by diluting DNA and RNA to appropriate concentrations if necessary.
- 2. Measure the absorbance of each sample at 260 nm and 280 nm.
- 3. Use the absorbance values to calculate the concentration of DNA and RNA.
- 4. Calculate the ratios of interest (sugar to phosphate ratios) using appropriate equations or standards.

Expected Results:

- The experiment is expected to yield values that reflect the relative proportions of sugar and phosphate in DNA and RNA molecules.
- Typically, DNA has a slightly higher sugar to phosphate ratio compared to RNA due to the difference in the sugars (deoxyribose vs. ribose).
- The results can be compared against theoretical values or known standards to validate the accuracy of the measurements.

This experiment provides essential information about the molecular composition of nucleic acids, which is fundamental for understanding their structure and biological functions.

Determination of melting temperature (Tm) of DNA

Aim:

The aim of this experiment is to determine the melting temperature (Tm) of DNA.

Principle:

The melting temperature (Tm) of DNA refers to the temperature at which half of the DNA molecules in a sample are in the double-stranded state and half are in the single-stranded state. It is influenced by factors such as DNA sequence, length, GC content, and ionic conditions. Tm can be determined by measuring the absorbance of DNA at different temperatures and plotting it to observe the transition from double-stranded to single-stranded DNA.

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Materials and Methodology:

Materials:

- DNA sample (e.g., PCR product, synthetic DNA)
- Spectrophotometer
- Temperature-controlled cuvette holder
- Buffer solution (appropriate for DNA stability, e.g., Tris-EDTA)
- Thermal cycler or water bath capable of controlled temperature increments
- Pipettes and tips
- Micro centrifuge tubes

Methodology:

1. Preparation of DNA Sample:

- Purify the DNA sample to remove contaminants that could affect Tm measurements.
- Quantify the concentration of DNA using a spectrophotometer or fluorometer.

2. Preparation of Buffers and Solutions:

- Prepare a buffer solution (e.g., Tris-EDTA) to stabilize DNA during measurements.
- 3. Setting Up the Experiment:
 - Set up the spectrophotometer with a temperature-controlled cuvette holder.

4. Measurement Protocol:

- Place the DNA sample in a cuvette and insert it into the spectrophotometer.
- Measure the absorbance of the DNA sample at a wavelength appropriate for DNA (e.g., 260 nm).
- \circ Record the baseline absorbance at an initial temperature.

5. Temperature Ramp:

• Gradually increase the temperature of the cuvette holder in small increments (e.g., 0.5°C per minute).

Lab manual for MSc

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 $\circ~$ Continuously measure the absorbance of the DNA sample as the temperature increases.

6. Data Collection:

- Plot the absorbance versus temperature to observe changes indicative of DNA denaturation.
- Identify the temperature at which the absorbance curve shows a significant decrease, indicating the Tm.

Procedure:

- 1. Prepare the DNA sample and buffer solution.
- 2. Set up the spectrophotometer and calibrate it if necessary.
- 3. Place the DNA sample in the cuvette and record the initial absorbance.
- 4. Start increasing the temperature gradually while recording absorbance at each temperature increment.
- 5. Continue until the absorbance curve indicates the DNA has denatured sufficiently.
- 6. Determine the temperature at the midpoint of the transition as the Tm.

Example Result:

After conducting the experiment and plotting the absorbance versus temperature, the following data was obtained:

- Initial absorbance at 260 nm: 0.8
- Absorbance decreased gradually with increasing temperature.
- Significant decrease in absorbance observed around 80°C.
- Tm determined to be approximately 82°C based on the midpoint of the denaturation curve.

This result indicates that the DNA sample starts to denature around 80°C and is fully denatured by 82°C, which represents the melting temperature (Tm) of the DNA under the experimental conditions used.